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## Continuous Approximation for Interaction Energy of Adamantane Encapsulated Inside Carbon Nanotubes

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Linear forms of crystalline diamond constitute a new one-dimensional nanomaterial. They assemble within carbon nanotubes which serve as a template and by taking diamondoids as building blocks. Template synthesis of linear chain nanodiamonds have been considered recently using diamantine polymers. The smallest building block of diamond is adamantane, which is the smallest unit with a diamond lattice. The interaction energy for two adjacent adamantane molecules and that of adamantane molecules encapsulated inside carbon nanotubes are investigated. The Lennard-Jones potential and the continuous approximation are utilized to derive analytical expressions for these interaction energies from a highly simplified model. The derived equilibrium distance is found to be within 3% of density functional calculations and an equilibrium distance of 3.281 Angstroms between two adamantane molecules is determined. The smallest carbon nanotube radius  $b_0$  that can encapsulate the adamantane molecule and the radius of the tube  $b_{max}$  that gives the maximum suction energy, are calculated to depend linearly on the adamantane radius. For larger diameter tubes, the off axis position is predicted, and the equilibrium distance between the molecule and the tube wall is found to be close to the interlayer spacing in graphene.

**Keywords:** Adamantane, carbon nanotube, Lennard-Jones potential, interaction energy.